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(54) Title: CHEMICAL RINSE COMPOSITION

(57) Abstract: The present invention relates to a thinner composition for removing resist used in TFT-LCD manufacturing processes, and more particularly to a thinner composition for removing resist that comprises : a) 0.1 to 5wt% of an inorganic alkali compound; b) 0.1 to 5wt% of an organic amine; c) 0.1 to 30wt% of an organic solvent; d) 0.01 to 5wt% of a surfactant comprising an ionic surfactant and a non-ionic surfactant in the weight ration of 1:5 to 1:25; and e) 60 to 99wt% of water. The thinner composition for removing resist of the present invention has good efficiency of removing unwanted resist film constituents formed on the edge of the resist film or at the back of the substrate in TFT-LCD device and semiconductor device manufacturing processes. Also, it does not have the problem of equipment corrosion.

TITLE OF THE INVENTION

## CHEMICAL RINSE COMPOSITION

BACKGROUND OF THE INVENTION**(a) Field of the Invention**

5       The present invention relates to a thinner composition for removing photosensitive resin (resist) film, used in TFT-LCD (thin film transistor liquid crystal display) and semiconductor device manufacturing processes, and more particularly to a thinner composition capable of effectively removing unnecessary film constituents formed at edges of the resist film coated on the substrate or formed at  
10      the back of the substrate during manufacturing processes of TFT-LCD devices and semiconductor devices.

**(b) Description of the Related Art**

In pattern formation of a fine circuit like a TFT-LCD circuit or a semiconductor integrated circuit, a resist composition is uniformly coated or applied  
15      on an insulation film or a conductive metal film formed on the substrate. Then, the coated resist composition is exposed and developed in the presence of a mask having a certain pattern to acquire the pattern. The metal film or the insulation film is etched using the patterned resist film as mask, and the remaining resist film is removed to form a fine circuit. When manufacturing TFT-LCDs or semiconductor  
20      devices with this lithography method, the resist film should be formed on a substrate such as glass or silicon wafer, and the substrate should be rinsed with a thinner prior

to exposure and developing of the resist film, in order to remove unnecessary resist constituents on the edge of the formed resist film and unnecessary resist film formed at the back of the substrate.

For thinners to rinse and remove the resist film on the substrate, water, 5 inorganic and organic alkaline thinners, and organic amine thinners like monoethanolamine are known. With the inorganic alkaline thinners, inorganic materials remain after removing unnecessary films, so that they may contaminate the processing equipment, and the removing efficiency is poor. Therefore, organic alkalis and organic amines are used as main components of the thinner, and 10 inorganic alkalis as minor components. The organic alkalis and organic amines leave few unwanted materials after they are evaporated, so that they do not corrode the equipment. Additionally, since they have good solubility to the resist, they have superior resist-removing efficiency.

Other thinner compositions comprise inorganic alkalis like potassium 15 hydroxide, sodium hydroxide, sodium phosphate, sodium silicate, sodium carbonate, or sodium hydrogen carbonate; organic alkalis like tetramethylammonium hydroxide; and organic solvents like dipropylene glycol monomethylether, propylene glycol monomethylether, propylene glycol monomethyl ether acetate, or n-butyl acetate. However, these organic solvents do not offer sufficient resist-removing efficiency, 20 and in particular, an increase in the content of inorganic alkalis causes problems with equipment corrosion after evaporation.

Therefore, development of a new thinner composition having superior resist-removing efficiency and that is capable of preventing equipment corrosion is needed.

### SUMMARY OF THE INVENTION

5       The present invention was made in consideration of the problems of the prior art, and it is an object of the present invention to provide a thinner composition for removing a resist in TFT-LCD device and semiconductor device manufacturing processes, having superior resist removing efficiency and that is capable of preventing equipment corrosion.

10       In order to achieve this object, the present invention provides a thinner composition for removing resist comprising: a) 0.1 to 5wt% of an inorganic alkali compound; b) 0.1 to 5wt% of an organic amine; c) 0.1 to 30wt% of an organic solvent; d) 0.01 to 5wt% of a surfactant comprising an anionic surfactant and a non-ionic surfactant in the weight ratio of 1:5 to 1:25; and e) 60 to 99wt% of water.

15       DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS

The present invention will now be explained in more detail.

The present inventors identified that a composition comprising one or more inorganic alkalis, an organic amine, one or more organic solvents, and a surfactant in a specific ratio has resist-removing efficiency and prevents equipment corrosion.

20       In a thinner composition for removing a resist of the present invention, the inorganic alkali (a) can be one or more compounds selected from potassium

hydroxide, sodium hydroxide, sodium phosphate, sodium silicate, sodium carbonate, and sodium hydrogen carbonate. The content of the inorganic alkali compound to the entire thinner composition is preferably 0.1 to 5wt%, and more preferably, 0.5 to 3wt%. If the content is below 0.5wt%, the resist-removing efficiency becomes poor  
5 because the composition does not penetrate into the resist well. Otherwise, if it exceeds 5wt%, equipment corrosion may arise after evaporation of the thinner composition, and development may be insufficient due to buildup of the thinner composition on the border of the resist removal area because of excessive penetration.

10 In the thinner composition for removing resist of the present invention, the organic amine (b) can be one or more compounds selected from monoethanolamine, diethanolamine, triethanolamine, monoethylamine, diethylamine, triethylamine, ethyleneglycol amine, propyleneglycol amine, butyleneglycol amine, diethyleneglycol amine, and dipropyleneglycol amine. The content of the organic amine to the entire  
15 thinner composition is preferably 0.1 to 5wt%, and more preferably 1 to 4wt%. If the content is below 0.1wt%, penetration into the resist polymer is weakened, so that the resist-removing rate is reduced. Otherwise, if it exceeds 5wt%, development may be insufficient due to buildup of the thinner composition on the resist-removal border because of excessive penetration.

20 Additionally, in the thinner composition for removing resist of the present invention, the organic solvent (c) is preferably a compound that is miscible with water

and that sufficiently dissolves the resist and organic amine compound. To be specific, one or more compounds selected from ethyleneglycol phenylether, propyleneglycol phenylether, butyleneglycol phenylether, diethyleneglycol phenylether, dipropyleneglycol phenylether, dipropyleneglycol monomethylether, 5 diethyleneglycol monoethyleneether, propyleneglycol monomethylether, propyleneglycol monomethylether acetate, N-methylpyrrolidone (NMP), N-ethyl pyrrolidone (NEP), N-propyl pyrrolidone (NPP), N-hydroxymethyl pyrrolidone, and N-hydroxyethyl pyrrolidone can be used. The content of the organic solvent is preferably 0.1 to 30wt%, and more preferably 1 to 10wt%. If the content is below 10 0.1wt%, the solubility to the resist and organic amine compound becomes poor. Otherwise, if it exceeds 30wt%, treatment of the waste solution becomes a problem.

In the thinner composition for removing resist of the present invention, the surfactant (d) is preferably a mixture of an anionic surfactant and a non-ionic surfactant. The surfactant should be miscible with water and soluble in the organic solvent. The anionic surfactant increases solubility to the organic solvent and water. 15 For the anionic surfactant, sodium lauryl sulfate, sodium alkyl sulfate, etc. can be used. For the non-ionic surfactant, polyoxyethyl ether, polyoxypropyl ether, polyoxyethyl octylphenyl ether, polyoxypropyl octylphenyl ether, polyoxyethyl propyl ether, polyoxyethylpropyl octylphenyl ether, and a mixture thereof can be used. The mixing ratio (weight ratio) of the anionic surfactant and non-ionic surfactant is 20 preferably 1:5 to 1:25. If the mixing ratio is below 1:5, solubility to the organic

solvent and water is poor. Otherwise, if it exceeds 1:25, the resist removing efficiency becomes poor. The content of the surfactant is preferably 0.01 to 5wt%. If the content is below 0.01wt%, mixing of the organic amine, organic solvent, and water is difficult. Otherwise, if it exceeds 5wt%, the resist-removing efficiency of the  
5 thinner composition becomes poor.

In the thinner composition for removing resist of the present invention, water (e), an essential constituent, is preferably pure water filtered through an ion-exchange resin, and more preferably ultrapure water whose specific resistance is larger than 18MΩ. The content of water to the entire thinner composition is  
10 preferably 60 to 99wt%. If the content is below 60wt%, treatment of the waste solution becomes a problem. Otherwise, if it exceeds 99wt%, the resist-removing efficiency becomes poor.

The thinner composition according to the present invention can solve the problems of insufficient rinsing efficiency and contamination of equipment of the  
15 conventional mixture of inorganic alkalis like potassium hydroxide, sodium hydroxide, sodium phosphate, sodium silicate, sodium carbonate, and sodium hydrogen carbonate; organic alkalis like tetramethylammonium hydroxide; and organic solvents like dipropylene glycol monomethylether, propylene glycol monomethylether, propylene glycol monomethylether acetate, and n-butyl acetate. In particular, it can  
20 remove even fine resist films.

Hereinafter, the present invention is described in more detail through

Examples. However, the following Examples are only for the understanding of the present invention, and the present invention is not limited to the following Examples. In the following Examples, the percentage and mixing ratio are based on weight, unless otherwise mentioned.

5 [Example]

Examples 1 to 8 and Comparative Examples 1 to 4

Thinner compositions of Examples 1 to 8 and Comparative Examples 1 to 4 were prepared by mixing organic amines, organic solvents, surfactants, and water, in the ratios shown in Table 1.

10 [Table 1]

| Contents<br>(wt%)          |                                 | Examples |     |     |     |     |     |     |     | Comparative<br>Examples |     |     |     |
|----------------------------|---------------------------------|----------|-----|-----|-----|-----|-----|-----|-----|-------------------------|-----|-----|-----|
|                            |                                 | 1        | 2   | 3   | 4   | 5   | 6   | 7   | 8   | 1                       | 2   | 3   | 4   |
| (a)<br>Inorganic<br>alkali | KOH                             | 0.5      | 0.5 | 0.5 | 1.0 | 1.0 | 1.0 | 0.5 | 0.5 | -                       | -   | -   | -   |
|                            | Na <sub>2</sub> CO <sub>3</sub> | 1.0      | 1.0 | 1.0 | 2.0 | 2.0 | 2.0 | 2.0 | 2.0 | -                       | -   | -   | -   |
|                            | NaOH                            | -        | -   | -   | -   | -   | -   | -   | -   | 0.3                     | 6.0 | -   | -   |
|                            | TMAH                            | -        | -   | -   | -   | -   | -   | -   | -   | -                       | -   | 0.6 | 0.6 |
| (b)<br>Organic<br>amine    | MEA                             | 0.5      | -   | -   | -   | -   | -   | 0.5 | -   | -                       | -   | -   | -   |
|                            | DEGA                            | -        | -   | -   | 0.5 | -   | -   | -   | -   | -                       | -   | -   | -   |
|                            | TEA                             | -        | 0.5 | -   | -   | 0.5 | -   | -   | 0.5 | -                       | -   | -   | -   |
|                            | TEOA                            | -        | -   | 0.5 | -   | -   | 0.5 | -   | -   | -                       | -   | -   | -   |

|                           |       |      |      |      |      |      |      |      |      |      |      |      |      |
|---------------------------|-------|------|------|------|------|------|------|------|------|------|------|------|------|
|                           | NMP   | 1.0  | 1.0  | 1.0  | -    | -    | 3.0  | 3.0  | 3.0  | -    | -    | -    | -    |
|                           | PPOH  | 1.0  | -    | -    | 1.0  | 1.0  | 1.0  | -    | -    | -    | -    | -    | -    |
| (c)<br>Organic<br>solvent | DPGME | -    | 3.0  | -    | 5.0  | -    | -    | 5.0  | -    | -    | 5.0  | -    | 5.0  |
|                           | PGMEA | -    | -    | 3.0  | -    | 5.0  | -    | -    | 5.0  | -    | -    | -    | -    |
|                           | nBA   | -    | -    | -    | -    | -    | -    | -    | -    | 5.0  | 5.0  | 5.0  | 5.0  |
|                           | PGME  | -    | -    | -    | -    | -    | -    | -    | -    | 5.0  | -    | 5.0  | -    |
| (d)<br>Surfactant         | POEO  | 1.0  | 1.0  | 1.0  | 1.0  | 2.0  | 2.0  | 2.0  | 2.0  | -    | -    | -    | -    |
|                           | ES    | 0.1  | 0.1  | 0.1  | 0.1  | 0.1  | 0.1  | 0.1  | 0.1  | -    | -    | -    | -    |
| (e) Water                 |       | 94.9 | 92.9 | 92.9 | 89.4 | 88.4 | 90.4 | 86.9 | 86.9 | 89.7 | 84.0 | 89.4 | 89.4 |

In Table 1, TMAH is tetramethylammonium hydroxide, MEA is monoethanolamine, DEGA is diethyleneglycol amine, TEA is triethylamine, TEOA is triethanolamine, NMP is N-methylpyrrolidone, PPOH is propyleneglycol phenylether, DPGME is dipropyleneglycol monomethyl ether, PGMEA is propyleneglycol monomethyl ether, POEO is polyoxyethyl octylphenyl ether, and ES is sodium lauryl sulfate.

#### Physical property test

Sample preparation: A commonly used color resist composition (FujiFilm 10 Arch's CR-8131L, CG-8130L, CB-8140L, CR-8100L, CG-8100L, and CB-8100L) was spin-coated on LCD glass substrates on which a Cr BM (black matrix) had been

deposited, so that the final film thickness becomes 1.0 to 2.0 $\mu\text{m}$ . Samples were prepared by vacuum-drying the substrates in a chamber (0.5 torr) for 60sec.

The glass substrates were dipped in the thinner compositions of Examples 1 to 8 and Comparative Examples 1 to 4 for 2 sec. After rinsing the substrates with deionized water, the removal status of unwanted films on the edge was observed with the naked eye and an optical electron microscope (LEICA's FTM-200). The results are shown in Table 2.

[Table 2]

| Classification  | Resist removing efficiency |                             |
|-----------------|----------------------------|-----------------------------|
|                 | Naked eye                  | Optical electron microscope |
| Example 1       | Good                       | Good                        |
| Example 2       | Good                       | Good                        |
| Example 3       | Good                       | Good                        |
| Example 4       | Good                       | Good                        |
| Example 5       | Good                       | Good                        |
| Example 6       | Good                       | Good                        |
| Example 7       | Good                       | Good                        |
| Example 8       | Good                       | Good                        |
| Comp. Example 1 | Poor                       | Poor                        |
| Comp. Example 2 | Good                       | Poor                        |
| Comp. Example 3 | Poor                       | Poor                        |
| Comp. Example 4 | Poor                       | Poor                        |

As shown Table 2, thinner compositions according to the present invention (Examples 1 to 8) showed good resist-removing efficiency. On the contrary, thinner compositions of Comparative Examples 1 to 4 showed poor resist-removing

efficiency.

As explained above, the thinner composition for removing resist of the present invention has good efficiency for removing unwanted resist film constituents formed on the edge of the resist film or at the back of the substrate in TFT-LCD device and semiconductor device manufacturing processes. Also, it does not have 5 the problem of equipment corrosion.

**WHAT IS CLAIMED IS:**

1. A thinner composition for removing resist, comprising:
  - a) 0.1 to 5wt% of an inorganic alkali compound;
  - b) 0.1 to 5wt% of an organic amine;
  - c) 0.1 to 30wt% of an organic solvent;
  - d) 0.01 to 5wt% of a surfactant comprising an anionic surfactant and a non-ionic surfactant in the weight ratio of 1:5 to 1:25; and
  - e) 60 to 99wt% of water.
2. The thinner composition for removing resist according to Claim 1, wherein  
10 the inorganic alkali compound is one or more compounds selected from potassium hydroxide, sodium hydroxide, sodium phosphate, sodium silicate, sodium carbonate, and sodium hydrogen carbonate.
3. The thinner composition for removing resist according to Claim 1, wherein  
the organic amine is one or more compounds selected from monoethanolamine,  
15 diethanolamine, triethanolamine, monoethylamine, diethylamine, triethylamine, ethyleneglycol amine, propyleneglycol amine, butyleneglycol amine, diethyleneglycol amine, and dipropyleneglycol amine.
4. The thinner composition for removing resist according to Claim 1, wherein  
the organic solvent is one or more compounds selected from ethyleneglycol  
20 phenylether, propyleneglycol phenylether, butyleneglycol phenylether,  
diethyleneglycol phenylether, dipropyleneglycol phenylether, N-methylpyrrolidone

(NMP), N-ethyl pyrrolidone (NEP), N-propyl pyrrolidone (NPP), N-hydroxymethyl pyrrolidone, and N-hydroxyethyl pyrrolidone.

5. The thinner composition for removing resist according to Claim 1, wherein the anionic surfactant is sodium lauryl sulfate or sodium alkyl sulfate.
6. The thinner composition for removing resist according to Claim 1, wherein the non-ionic surfactant is one or more compounds selected from polyoxyethyl ether, polyoxypropyl ether, polyoxyethyl octylphenyl ether, polyoxypropyl octylphenyl ether, polyoxyethyl propyl ether, and polyoxyethylpropyl octylphenyl ether.

# INTERNATIONAL SEARCH REPORT

International application No.

PCT/KR02/02117

## A. CLASSIFICATION OF SUBJECT MATTER

**IPC7 G03F 7/32**

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC7 G03F 7/32, G03F 7/42, G03F 7/38, G03F 7/40, H01L 21/027, B08B 3/04, C11D 3/20

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched  
KR, JP: Classes as above

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)  
PAJ, NPS, ESPACENET, USPTO

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

| Category* | Citation of document, with indication, where appropriate, of the relevant passages | Relevant to claim No. |
|-----------|--|-----------------------|
| X         | US 5,962,197 A (ANALYZE INC.) 05. Oct. 1999<br>See the whole document              | 1 - 6                 |
| X         | US 6,274,296 B1 (SHIPLEY COMPANY,L.L.C.,) 14. Aug. 2001<br>See the whole document  | 1 - 6                 |
| P, X      | US 6,379,875 B2 (SHIPLEY COMPANY,L.L.C.,) 30. Apr. 2002<br>See the whole document  | 1 - 6                 |
| P, X      | KR 2002-63096 A (DONGWOO FINECHEM.) 01. Aug. 2002<br>See the whole document        | 1 - 6                 |
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| A         | JP 08-87118 A (TOKYOOHKAKOGYO CO LTD) 02. Apr. 1996<br>See the whole document      | 1 - 6                 |

Further documents are listed in the continuation of Box C.

See patent family annex.

### \* Special categories of cited documents:

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**INTERNATIONAL SEARCH REPORT**

Information on patent family members

International application No.

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| Patent document cited in search report | Publication date | Patent family member(s) | Publication date |
|--|------------------|-------------------------|------------------|
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| KR 2002-63096 A                        | 01. Aug. 2002    | none                    |                  |
| KR 2003-359 A                          | 06. Jan. 2003    | none                    |                  |
| JP 63-231343 A                         | 27. Sep. 1988    | none                    |                  |
| JP 08-87118 A                          | 02. Apr. 1996    | none                    |                  |

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